

Phosphoenolpyruvic Acid, an Intermediate of Glycolysis, Attenuates Cellular Injury Induced by Hydrogen Peroxide and 2-Deoxy-D-glucose in the Porcine Proximal Kidney Tubular Cell Line, LLC-PK₁

Yuki Kondo,^a Yoichi Ishitsuka,^{*,a}
Daisuke Kadowaki,^{b,c} Minako Nagatome,^a
Yusuke Saisho,^a Masataka Kuroda,^a
Sumio Hirata,^c Mitsuru Irikura,^a
Naotaka Hamasaki,^d and Tetsumi Irie^{a,c}

^aDepartment of Clinical Chemistry and Informatics, Graduate School of Pharmaceutical Sciences, Kumamoto University, 5-1 Oe-honmachi, Kumamoto 862-0973, Japan, ^bDepartment of Biopharmaceutics, Graduate School of Pharmaceutical Sciences, Kumamoto University, 5-1 Oe-honmachi, Kumamoto 862-0973, Japan, ^cCenter for Clinical Pharmaceutical Sciences, Faculty of Pharmaceutical Sciences, Kumamoto University, 5-1 Oe-honmachi, Kumamoto 862-0973, Japan and ^dDepartment of Clinical Chemistry and Laboratory Medicine, Nagasaki International University, Sasebo 859-3298, Japan

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This study was conducted to clarify whether phosphoenolpyruvate (PEP), an intermediate substance of glycolysis, has the potential to attenuate cellular injury induced by oxidative stress or dysfunctions in energy metabolism *in vitro*. PEP (0.5–10 mM) attenuated hydrogen peroxide (H₂O₂)-induced cellular injury in the porcine proximal kidney tubular cell line, LLC-PK₁ in a dose-dependent manner. PEP also prevented cellular injury in LLC-PK₁ cells induced by the glycolysis inhibitor, 2-deoxy-D-glucose (2-DG). In addition, PEP significantly enhanced the degradation of H₂O₂. The prevention of H₂O₂-induced cellular injury mediated by PEP was more potent than that of the carbohydrates, glucose and trehalose, which are used as components of organ preservation solutions for clinical transplantation. In conclusion, we demonstrated that PEP is a bifunctional carbohy-

drate with anti-oxidant properties and suggest that PEP is potentially useful as an organ preservation agent in clinical transplantation.

Key words—Phosphoenolpyruvate, carbohydrate, oxidative stress, transplantation, cellular injury

INTRODUCTION

Ischemia/reperfusion injury, which results from the interruption of blood flow and subsequent organ reperfusion, contributes to graft dysfunction and acute graft rejection; it is, therefore, a major problem in clinical organ transplantation.^{1–3} Oxidative stress and compromised energy metabolism are associated, at least in part, with ischemia/reperfusion injury.^{4–7} To inhibit this, organ preservation solutions such as Euro-Collins solution and University of Wisconsin solution are used.^{8–10} These solutions are generally composed of electrolytes and buffers, with carbohydrates such as glucose and trehalose added to prevent energy loss. In addition, to inhibit the generation of reactive oxygen species during cold preservation and ischemia/reperfusion, the anti-oxidant glutathione (GSH) is added to the University of Wisconsin solution.^{11–13} Although they are essential for clinical transplantation, there are several disadvantages to these solutions, including chemical instability and high viscosity. Also, they are not cost-effective for short-term storage.¹⁴ Therefore, a new, effective reagent for organ preservation is critical.

Phosphoenolpyruvate (PEP) is an intermediate of glycolysis and gluconeogenesis that contains a high-energy phosphate bond. PEP can penetrate the cell membrane and transfer its high-energy phosphate group to adenosine 5'-diphosphate (ADP) to replenish intracellular ATP levels.^{15,16} It has been reported that PEP improves the energy status of the heart^{17,18} and skeletal muscle¹⁹ after ischemia, and of the liver after ischemia-reperfusion.¹⁵ Hojo *et al.*²⁰ demonstrated that the survival of transplanted kidney grafts was improved by Euro-Collins solution containing 10 mM PEP in a canine experimental kidney transplantation model. However, little is known about the precise mechanism underlying the protective effects of PEP and there is no evidence that PEP has the potential to attenuate cellular injury induced by oxidative stress, or dysfunctions

*To whom correspondence should be addressed: Department of Clinical Chemistry and Informatics, Graduate School of Pharmaceutical Sciences, Kumamoto University, 5-1 Oe-honmachi, Kumamoto 862-0973, Japan. Tel. & Fax: +81-96-371-4559; E-mail: y-zuka@gpo.kumamoto-u.ac.jp

in energy metabolism, which play important roles in ischemia/reperfusion injury.

Based on these facts, we designed an *in vitro* experiment to examine the effects of PEP on cellular injury induced by hydrogen peroxide (H_2O_2) or the glycolysis inhibitor, 2-deoxy-D-glucose (2-DG), in the porcine proximal kidney tubular cell line, LLC-PK₁. To evaluate the usefulness of PEP as an organ preservation agent, we compared the anti-oxidative effects of PEP with that of other components used in organ preservation solutions, such as glucose, trehalose and GSH.

MATERIALS AND METHODS

Materials — Sodium phosphoenolpyruvate monohydrate was kindly donated by Ube Kousan (Yamaguchi, Japan). H_2O_2 and 2-DG were purchased from Wako Pure Chemicals (Osaka, Japan). The Cell Counting Kit was obtained from Dojindo Laboratories (Kumamoto, Japan). Medium 199 Earle's (M199) was obtained from Invitrogen Japan (Tokyo, Japan). Fetal bovine serum (FBS) was purchased from Thermo Scientific HyClone (Logan, UT, U.S.A.). All other reagents and solvents were of reagent grade. De-ionized and distilled bio-pure grade water was used throughout the study.

Cell Culture — The LLC-PK₁ cell line was obtained from the Japanese Cancer Research Resources Bank (Tokyo, Japan). LLC-PK₁ cells were cultured in M199 supplemented with 3% FBS, 100 IU/ml of penicillin, 100 μ g/ml streptomycin at 37°C and 5% CO₂. Confluent cultures were washed with phosphate-buffered saline (PBS) pH 7.4, detached with EDTA and trypsin, centrifuged and subcultured in 96-well plates. Cells were seeded at a density of 1×10^4 cells/well and grown on culture dishes for 24 hr before use in the cell viability experiments.

Cell Viability Assays — To evaluate the viability of LLC-PK₁ cells, mitochondrial dehydrogenase activity was measured using a modified MTT assay (the WST-1 assay) and the Cell Counting Kit according to the manufacturer's protocol. Cells were incubated with either H_2O_2 or 2-DG at 37°C and 5% CO₂ and the cell viability was assessed after 24 hr. In a preliminary study, we determined the minimum concentrations of H_2O_2 and 2-DG that induced a decrease in cell viability with high reproducibility and optimal concentrations were chosen for treatment. Based on the results, we used H_2O_2 or 2-DG at con-

centrations of 0.1 mM and 15 mM, respectively. To evaluate the effects of PEP on cell viability, PEP was used at doses up to 10 mM, as the maximum non-cytotoxic dose in LLC-PK₁ cells was 20 mM (data not shown).

Measurement of Hydrogen Peroxide Decomposition — We prepared a reaction mixture containing: 0.1 mM H_2O_2 , 120 mM KCl and 50 mM Tris-HCl, pH 7.4 and then added PEP (0.1–10 mM) to the mixture. The reaction was allowed to proceed for 30 min at 37.8°C and terminated by the addition of a stop solution (25 mg/ml potassium biphthalate, 2.5 mg/ml NaOH, 82.5 mg/ml potassium iodide, and 0.25 mg/ml ammonium molybdate). The absorbance of the mixture was measured at 350 nm in a spectrophotometer (V-530; Jasco, Tokyo, Japan) and the concentration of H_2O_2 was determined by interpolation of the data using a standard curve.

Statistical Analysis — Results were expressed as mean \pm S.D. Multiple comparisons were made to examine the statistical significance of the data. When uniform variance was identified by Bartlett's analysis ($p < 0.05$), one-way analysis of variance (ANOVA) was used to test for statistical significance. When significant differences ($p < 0.05$) were identified the data were further analyzed using Tukey's multiple range test.

RESULTS

Effects of PEP Treatment on the Decrease in Cell Viability Induced by H_2O_2 or 2-DG in LLC-PK₁ Cells

LLC-PK₁ cells treated with 0.1 mM H_2O_2 for 24 hr showed a 67% decrease in cell viability compared with the control (Fig. 1A). When 0.5–10 mM PEP was added to the culture medium just after H_2O_2 treatment, the decrease in cell viability induced by H_2O_2 was significantly attenuated in a dose-dependent manner (Fig. 1A).

As shown in Fig. 1B, treatment with 15 mM 2-DG for 24 hr significantly reduced cell viability (to approximately 60% of the value of the control group). This decrease in viability was significantly attenuated by PEP (10 mM, to approximately 87% of the value of the control group).

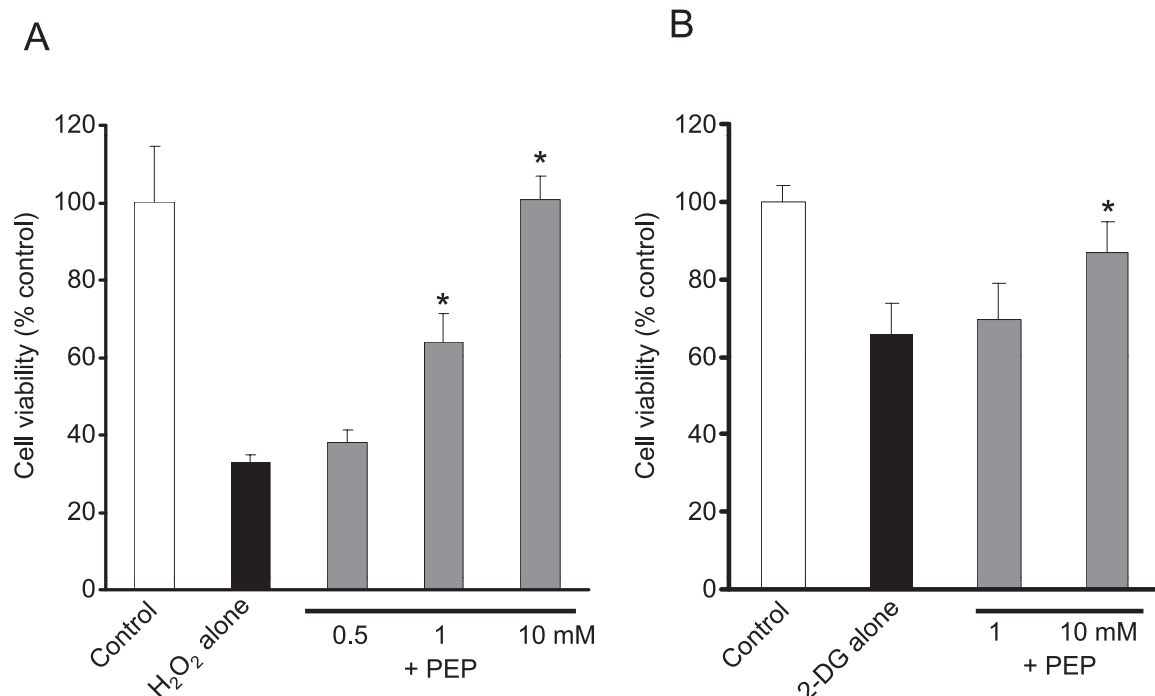


Fig. 1. PEP Abrogates the Decrease in Cell Viability Induced by H₂O₂ or 2-DG in LLC-PK₁ Cells

LLC-PK₁ cells exposed to 0.1 mM H₂O₂ (A) or 15 mM 2-DG (B) in either the presence or absence of PEP at different concentrations (0.5–10 mM). Cell viability was measured 24 hr after the addition of H₂O₂ or 2-DG using the WST-1 assay. Data are expressed as percent cell survival relative to the control group. Each bar represents mean \pm S.D. ($n = 8$). * $p < 0.01$ compared with the H₂O₂-treated or 2-DG-treated group.

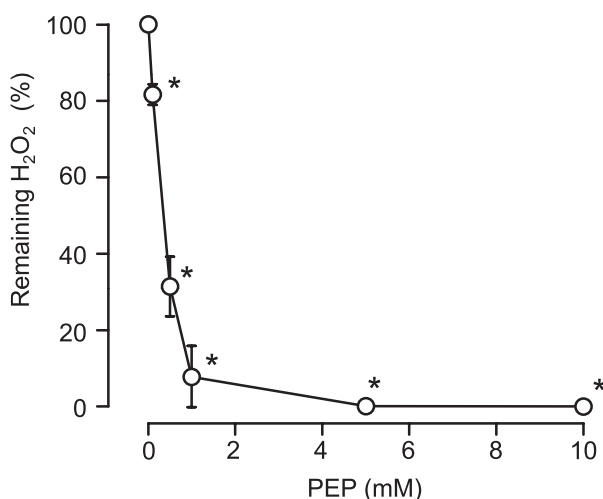


Fig. 2. Effects of PEP on H₂O₂ Decomposition

Data are expressed as percent H₂O₂ relative to the PEP-free control. Each point represents mean \pm S.D. ($n = 4$). * $p < 0.01$.

Evaluation of PEP-mediated H₂O₂ Decomposition

As shown in Fig. 2, treatment with PEP significantly reduced the H₂O₂ content of the reaction mixture, which was completely abolished by PEP

at concentrations ≥ 5 mM. To compare the ability of PEP to degrade H₂O₂ with that of the components of Euro-Collins or University of Wisconsin organ preservation solutions (such as glucose, trehalose and GSH) we determined the concentration of each reagent required to yield 50% decomposition of 0.1 mM H₂O₂ (50% effective concentration, EC₅₀). The EC₅₀ value for PEP was 0.26 mM compared with 200 and 120 mM (close to the concentrations used for organ preservation) for glucose and trehalose, respectively. The EC₅₀ of GSH and N-acetylcysteine (NAC), also used as antioxidants for organ preservation, were 0.07 and 0.02 mM, respectively.

Effects of PEP and Other Components in Organ Preservation Solutions on H₂O₂-induced Cellular Injury in LLC-PK₁ Cells

Glucose (194 mM) and trehalose (120 mM, the concentrations used in organ preservation solutions) had little effect in preventing the H₂O₂-mediated decrease in cell viability (Fig. 3). However, this decrease in viability was significantly abrogated by PEP (10 mM), GSH (3 mM) and NAC (10 mM, Fig. 3).

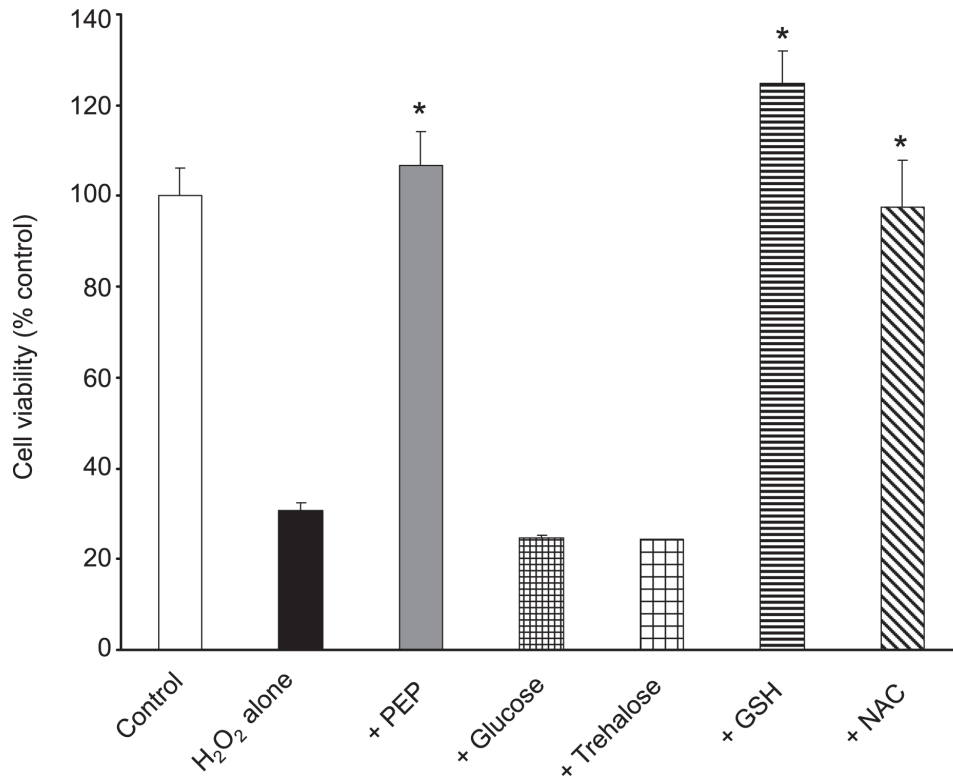


Fig. 3. Effects of PEP or Other Components of Organ Preservation Solutions on the Decrease in Cell Viability Induced by H₂O₂

LLC-PK₁ cells exposed to H₂O₂ in either the presence or absence of PEP (10 mM), glucose (194 mM), trehalose (120 mM), glutathione (3 mM) or NAC (10 mM). Cell viability was estimated 24 hr after the addition of H₂O₂. Data are expressed as percent cell survival relative to the control group. Each bar represents mean ± S.D. (*n* = 6). **p* < 0.01 compared with the H₂O₂-treated group.

DISCUSSION

In this study, we showed that PEP, an intermediate of glycolysis, protected LLC-PK₁ cells from injury induced by H₂O₂ and 2-DG. PEP was also able to mediate the decomposition of H₂O₂. These results indicate that PEP protects against cellular injury induced by oxidative stress and against dysfunctions in carbohydrate metabolism through, at least in part, an antioxidant effect.

Ischemia/reperfusion of allografts triggers oxidative stress leading to dysfunction and acute rejection.²⁾ The xanthine oxidase system is activated during ischemia/reperfusion and increased levels of H₂O₂ and hydroxyl radicals (decomposition products of H₂O₂) can injure the allograft; thus, H₂O₂-induced renal cell injury has frequently been used as an *in vitro* model of ischemia/reperfusion injury in the kidney.^{21,22)} The results of our current *in vitro* study support those from *in vivo* studies showing that PEP improves energy status and improves the function and survival of allografts.²⁰⁾ Our results also indicate that PEP protects allografts from acute

rejection, not only through improved ATP levels, but also via its antioxidant effects.

PEP contains a high-energy phosphate group and efficiently supplies energy to ischemic/reperfused organs *in vivo* and *in situ*.^{15,17–20)} Our current results suggest that the *in vitro* protective effects of PEP against cellular injury induced by the glycolysis inhibitor, 2-DG, used as an *in vitro* model of ischemic organ injury due to energy metabolism dysfunction,²¹⁾ may be due to the fact that PEP replenishes cellular ATP and acts as a cytoprotective agent against energy metabolism dysfunctions.

The protective effects of PEP against cellular injury mediated by oxidative stress and dysfunctions in carbohydrate metabolism protect allografts and organs from ischemia/reperfusion injury. Antioxidants are frequently used in organ preservation solutions to protect tissues from oxidative stress induced during preservation and ischemia/reperfusion.^{8,12)} University of Wisconsin solution contains glutathione and allopurinol, whereas ET-Kyoto solution contains NAC. These preservation solutions

seem to achieve better results than other preservation solutions.^{23,24)} In addition, carbohydrates such as glucose and trehalose are added to preservation solutions to reduce energy depletion from organs/cells during preservation. PEP may be an attractive bifunctional carbohydrate with the potential to act as both energy supplier and antioxidant. To establish the efficacy of PEP as a component of organ preservation solutions, it is important to identify its effects on other causes of allograft dysfunction, such as cold preservation and rewarming of the cells and organs and immune responses/inflammatory reactions.²⁵⁾ In addition, it is important to clarify whether PEP-containing organ preservation solutions are safe for both the allograft and the recipient. Further studies are therefore warranted.

In summary, we have shown that PEP protected LLC-PK₁ cells against injury induced by H₂O₂ and 2-DG and facilitated the decomposition of H₂O₂. The results suggest that PEP is a bifunctional carbohydrate with antioxidant effects and has potential for use as an agent for organ preservation in clinical transplantation.

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