

Preventive Effect of Preinduction of Metallothionein on Mutagenicity Caused by Benzo[*a*]pyrene

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The effect of pretreatment with zinc (Zn) compounds on the mutagenicity of benzo[*a*]pyrene (B[*a*]P) was investigated using metallothionein (MT)-I/II null mice. MT-I/II null mice and wild-type mice were subcutaneously administered ZnSO₄ once a day for 2 days and gavaged B[*a*]P at 24 hr after the last injection of ZnSO₄. B[*a*]P-induced micronucleus frequencies were reduced by Zn pretreatment in the wild-type mice but not in the MT-I/II null mice. Zn administration significantly increased the concentration of MT in the liver and bone marrow cells of wild-type mice, but the statuses of other cellular antioxidants, such as glutathione, catalase and superoxide dismutase, were unchanged. In addition, the activity of a major B[*a*]P metabolic activation enzyme, cytochrome P450 1A, was unchanged by Zn treatment in both MT-I/II null mice and wild-type mice. These results suggest that Zn pretreatment protects against the mutagenicity of B[*a*]P through the induction of MT synthesis. The amount of MT produced in animals may determine their sensitivity to B[*a*]P exposure.

Key words — metallothionein, benzo[*a*]pyrene, zinc, micronucleated reticulocyte

INTRODUCTION

Metallothionein (MT) is a low-molecular-weight protein in which cysteine residues comprise approximately 30% of the amino acids. It also has a high affinity for heavy metal ions such as zinc (Zn), copper, cadmium and mercury.¹⁾ There are four isoforms of MT, and MT-I and MT-II, the major isoforms, are known to be involved in many physiological and pathophysiological processes, such as intracellular storage, transport and metabolism of heavy metals in order to regulate essential trace metal homeostasis and protect against heavy metal toxicity.²⁾ It has also been suggested that MT reacts with free radicals and alkylating agents.^{3,4)} The synthesis of MT is induced by various factors, such as metals, hormones, cytokines, stress and chemicals.⁵⁾ Zn is known to be an effective inducer of MT synthesis, and MT primarily binds to Zn under physiological conditions.⁶⁾

Several studies have reported that pretreatment with Zn as an MT-inducer can depress the toxicities of cisplatin, adriamycin and cadmium.^{7,8)} Moreover, previous studies have shown that dietary Zn inhibits chemical carcinogenesis.^{9,10)} In these studies, however, it was unclear whether MT pre-induced by Zn directly prevents these toxicities, including chemical carcinogenesis, because Zn does not exclusively induce MT.

Benzo[*a*]pyrene (B[*a*]P) is one of the polycyclic aromatic hydrocarbons (PAHs) classified as carcinogenic to humans by International Agency for Research on Cancer (IARC).¹¹⁾ B[*a*]P has been shown to induce gene mutations, chromosomal aberrations and other types of genotoxic effects in various species and tissues.¹²⁾ Most humans are more or less constantly exposed to B[*a*]P, mainly through tobacco smoke, diesel engine exhaust fumes, various types of processed foods, coal-tar, creosotes and coke ovens.¹³⁾ B[*a*]P itself is chemically inert, and requires metabolic activation by cytochrome P450 (CYP) enzymes to exhibit carcinogenicity in animals and humans.¹⁴⁾ B[*a*]P is metabolized to 7,8-dihydroxy-9,10-epoxy-7,8,9,10-tetrahydrobenzo[*a*]pyrene (BPDE), the most car-

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cinogenic form of B[a]P, mainly by CYP1A1, 1A2 and 1B1, and binds to the exocyclic nitrogen of deoxyguanosine in DNA.¹⁵⁾ Moreover, during metabolic processes, B[a]P also produces quinone derivatives, which easily generate reactive oxygen species (ROS) and oxidatively damage DNA.¹⁶⁾ Our recent studies using MT-I/II null mice have shown that MT plays a protective role in B[a]P-induced DNA damage.¹⁷⁾ MT-I/II null mice, which are MT-I and MT-II double knockout mice, have been used to directly evaluate the biological mechanism involved.

In the present study using MT-I/II null mice and wild-type mice, we investigate the effect of preinduction of MT, using a Zn compound as an MT inducer, on micronuclei (MN) caused by B[a]P. We also investigate the effect of Zn treatment on the amounts of MT and glutathione (GSH) produced and the activities of catalase, superoxide dismutase (SOD) and CYP1A in the liver and bone marrow cells of MT-I/II null mice and wild-type mice.

MATERIALS AND METHODS

Chemicals— B[a]P was purchased from Sigma Chemicals Co. (St. Louis, MO, U.S.A.). Zinc sulfate (ZnSO₄), trichloroacetic acid, 5,5'-dithiobis(2-nitrobenzoic acid) (DTNB) and hydrogen peroxide (H₂O₂) were purchased from Wako Pure Chemical Industries, Ltd. (Osaka, Japan). SOD Assay kit-WST was purchased from Dojindo (Kumamoto, Japan).

Animals— MT-I/II null mice were kindly provided by Dr. K. H. A. Choo (Murdoch Institute for Research into Birth Defects, Royal Children's Hospital, Melbourne, Australia) and were of a mixed genetic background of 129 Ola and C57BL/6 strains.¹⁸⁾ F1 hybrid mice were mated with C57BL/6J mice and their offspring were backcrossed to C57BL/6J for six generations. Both MT-I/II null mice and wild-type mice were generated by mating of heterozygous mice.

MT-I/II null mice and wild-type mice were routinely bred in the vivarium of Gifu Pharmaceutical University. Both strains of mice were housed in cages in ventilated animal rooms with a controlled temperature of 24 ± 2°C, a relative humidity of 55 ± 10% and a 12 hr light/dark cycle. The mice were maintained on standard laboratory chow and tap water *ad libitum*, and received humane care throughout the experiment according to the guide-

lines of Gifu Pharmaceutical University.

Treatments— Ten-week-old male MT-I/II null mice and wild-type mice were randomized into control and experimental groups (5 mice for each group). The groups of mice were subcutaneously administered ZnSO₄ (100 μmol/kg) once a day for 2 days and orally administered B[a]P (250 mg/kg) once at 24 hr after the last administration of ZnSO₄. Peripheral blood was collected from the mouse tail vein under diethyl ether anesthesia at 48 hr after the B[a]P injection. In addition, the liver and bone marrow cells were removed from each mouse under diethyl ether anesthesia at 24 hr after the last administration of ZnSO₄.

MN Test— Peripheral blood smeared on glass slides (Matsunami Glass Industries, Osaka, Japan) was stained with acridine orange according to the method described by Hayashi *et al.*¹⁹⁾ For each animal, 1000 reticulocytes (RET) were scored for MN and the frequencies of micronucleated reticulocytes (MNRET) were expressed as a percentage of RET.

Measurement— MT (MT-I and MT-II isoforms) concentrations in the liver and bone marrow cells were measured by radioimmunoassay²⁰⁾ as modified by Nishimura *et al.*²¹⁾ using sheep anti-rat MT-I antiserum. 10% liver homogenate was prepared in 10 mM Tris-HCl. Protein was identified using BCA Protein Assay Kit (Thermo, Waltham, MA, U.S.A.) with bovine serum albumin as a standard.

GSH concentrations in the liver and bone marrow cells were measured by the method of Moron *et al.*²²⁾ The tissues were homogenized in phosphate buffered saline (PBS) to obtain a final concentration of 10% homogenate. The rate of reduction of DTNB was recorded spectrophotometrically at a wavelength of 412 nm. The GSH content was quantified by comparison with a standard curve generated with known amounts of GSH (reduced form) and expressed as μmol per mg protein.

For the measurement of catalase activity, we used the method of Aebi *et al.*²³⁾ One unit of catalase activity is equal to the amount of enzyme required to convert 1 μmol of H₂O₂ to oxygen and water in 1 min, and the activity of catalase was expressed as units of catalase/μg of protein.

The assay for SOD activity was measured using SOD Assay kit-WST according to the manufacturer's instructions. The tissues were homogenized in sucrose buffer to obtain a final concentration of 20% homogenate. One unit of SOD was defined as the enzyme activity causing 50% inhibition of the oxidation of water soluble tetrazolium salts (WST)-

1 reduction rate. The activity of SOD was expressed as units of SOD/mg of protein.

For the measurement of CYP1A activity, 7-ethoxyresorufin *O*-deethylation (EROD) activity in liver microsomes was determined by the method of Shimada *et al.*²⁴⁾

Statistical Analysis— All values were expressed as the mean \pm S.D. Differences in the mean values were assessed by analysis of variance (ANOVA) followed by Newman-Keuls tests for *post hoc* comparison. Differences were considered statistically significant at $p < 0.05$.

RESULTS

MN Test

The results of the MN test are shown in Fig. 1. No significant changes in MN production were found after treatment with ZnSO₄. The frequencies of MNRET were significantly increased by the administration of B[a]P in both the MT-I/II null mice and the wild-type mice; however, the frequency of MNRET in the MT-I/II null mice was significantly higher. Pretreatment with ZnSO₄ significantly decreased the frequency of MNRET increased by B[a]P in the wild-type mice but not in the MT-I/II null mice.

MT Concentration

MT concentrations were determined in the liver and bone marrow cells of MT-I/II null mice and wild-type mice at 24 hr after the last injection of ZnSO₄ (Table 1). In the wild-type mice, MT concentrations in the liver and bone marrow cells were increased by ZnSO₄ administration to 14- and 2.7-fold that of vehicle control mice, respectively. In contrast, MT concentrations in the liver and bone

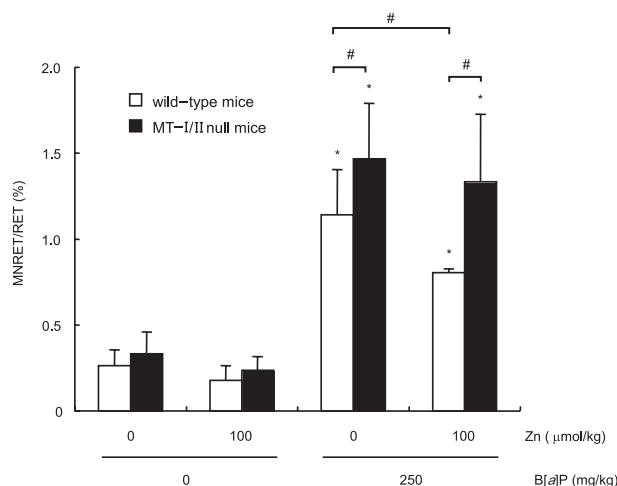


Fig. 1. Frequencies of MNRET in Peripheral Blood of MT-I/II Null Mice and Wild-type Mice Treated with B[a]P

Values are represented as mean \pm S.D. *Statistically significant differences from the corresponding control group ($p < 0.05$). Statistically significant differences are indicated with # $p < 0.05$.

Table 1. Concentrations of MT and GSH, and Activities of Catalase and SOD in Liver (A) and Bone Marrow Cells (B) of MT-I/II Null Mice and Wild-type Mice Pretreated with ZnSO₄

(A) Liver

	wild-type mice		MT-I/II null mice	
	saline	ZnSO ₄	saline	ZnSO ₄
MT	5.48 \pm 2.75	444.61 \pm 194.08*	N.D.	N.D.
GSH	0.33 \pm 0.05	0.31 \pm 0.02	0.37 \pm 0.04	0.35 \pm 0.03
catalase	41.43 \pm 16.87	43.44 \pm 7.59	39.91 \pm 5.08	42.50 \pm 5.66
SOD	128.99 \pm 13.71	139.37 \pm 13.07	129.37 \pm 9.01	131.35 \pm 9.62

MT, μ g/g tissue; GSH, μ mol/mg protein; catalase, IU/ μ g protein; SOD, U/mg protein. Values are means \pm S.D. Statistically significant differences from the control group. * $p < 0.05$. N.D.: not detected. The limit of detection for the MT assay is 1.85 μ g/g tissue.

(B) Bone marrow cells

	wild-type mice		MT-I/II null mice	
	saline	ZnSO ₄	saline	ZnSO ₄
MT	68.48 \pm 9.90	181.60 \pm 88.28*	N.D.	N.D.
GSH	262.58 \pm 50.75	320.25 \pm 92.64	270.39 \pm 54.20	266.52 \pm 57.1
catalase	89.13 \pm 11.95	86.79 \pm 10.79	82.05 \pm 15.23	71.65 \pm 18.06
SOD	73.03 \pm 22.82	69.52 \pm 26.09	94.24 \pm 28.21	65.70 \pm 11.85

MT, μ g/mg protein; GSH, μ mol/mg protein; catalase, IU/ μ g protein; SOD, U/mg protein. Values are means \pm S.D. Statistically significant differences from the control group. * $p < 0.05$. N.D.: not detected. The limit of detection for the MT assay is 50 μ g/mg protein.

Table 2. EROD Activities in the Liver and Bone Marrow Cells of MT-I/II Null Mice and Wild-type Mice Pretreated with ZnSO₄

	wild-type mice		MT-I/II null mice	
	saline	ZnSO ₄	saline	ZnSO ₄
liver	7.25 ± 1.00	6.35 ± 3.16	6.98 ± 1.48	7.54 ± 3.93
bone marrow cells	N.D.	N.D.	N.D.	N.D.

EROD, nmol/min per mg protein. Values are means ± S.D. N.D.: not detected.

marrow cells of MT-I/II null mice were below the limit of detection and were thus not induced by ZnSO₄.

GSH Concentration

GSH levels in the liver and bone marrow cells of both MT-I/II null mice and wild-type mice administered ZnSO₄ are shown in Table 1. No significant changes in GSH levels were observed between the ZnSO₄ treatment group and the vehicle controls in the tissues of both MT-I/II null mice and wild-type mice.

Antioxidant Enzymatic Activities

Table 1 also shows the catalase and SOD activities in the liver and bone marrow cells of both MT-I/II null mice and wild-type mice administered ZnSO₄. No significant changes in catalase and SOD activities were observed between the ZnSO₄ treatment group and the vehicle controls in the tissues of both MT-I/II null mice and wild-type mice.

CYP1A Enzymatic Activity

Table 2 shows the EROD activities in the liver and bone marrow cells of both MT-I/II null mice and wild-type mice administered ZnSO₄. Hepatic EROD activities were not changed between the ZnSO₄ treatment group and the untreated control group of both MT-I/II null mice and wild-type mice. The EROD activities of bone marrow cells were not detected in either MT-I/II null mice or wild-type mice, and were thus considered not to be induced by ZnSO₄ treatment.

DISCUSSION

We previously reported that MT-I/II null mice were more susceptible to B[a]P-induced chromosomal aberrations than wild-type mice.¹⁷⁾ In the present study, B[a]P-induced MN frequency was reduced by Zn pretreatment in wild-type mice but not in MT-I/II null mice. Our results suggest that basal

and metal-induced MT can play a protective role against the mutagenicity caused by B[a]P.

B[a]P must be metabolically activated to exert its toxicity, and CYP-dependent metabolic activation appears to be required. Nagalingam *et al.* reported an *in vitro* study that showed that BPDE, the most important metabolite, forms DNA adducts to act as a mutagen.²⁵⁾ CYP1A1, CYP1A2 and CYP1B1 have been shown to be the major enzymes in the metabolism of B[a]P.²⁶⁾ Pathways for the activation of B[a]P to its most carcinogenic metabolites by drug-metabolizing enzymes have extensively been studied.²⁷⁾ Several researchers have demonstrated clearly that CYP1A1 is involved in the metabolic activation of B[a]P into reactive intermediates capable of binding to DNA.²⁶⁾ Numerous reports have shown that one or more of the reactive intermediates, rather than the nonmetabolized parent compound, is responsible for B[a]P-mediated mutations and carcinogenesis.^{12, 28)} Expression of CYP1A1 is constitutively nil but is markedly induced in a large number of tissues after induction by PAHs.²⁹⁾ In addition, constitutive CYP1B1 is extremely low in the liver but is detectable after PAH exposure.²⁹⁾ In contrast, CYP1A2 is constitutively high in the mammalian liver. CYP1A1 and CYP1B1 are expressed in organs apart from the liver but CYP1A2 is mainly expressed in the liver. Our previous study showed that the oral administration of B[a]P increased EROD activities in both MT-I/II null mice and wild-type mice, as a result of CYP1A activity.¹⁷⁾ In the present study, we measured EROD activities in the liver and bone marrow cells, and found that they were not changed between the Zn treatment group and the untreated control group in both MT-I/II null mice and wild-type mice. Thus, Zn pretreatment does not affect B[a]P metabolic activation.

Metal compounds containing metals such as cadmium, Zn and bismuth can induce MT synthesis *in vivo* and *in vitro*.³⁰⁾ Several investigators have reported that metal compounds induced MT to protect against the cytotoxicity and organ toxicity

of various chemicals.³¹⁾ For example, pretreatment with bismuth compounds, as MT inducers, suppressed the MN induction by chromium compounds and selenium compounds.³²⁾ Anticancer drugs, such as cisplatin, adriamycin, cyclophosphamide and L-phenylalanine mustard, induced MN, but pretreatment with MT inducers could prevent this.³³⁾ Recently, a few reports have described using MT-I/II null mice to demonstrate that the protective effects of Zn pretreatment on cisplatin-induced renal toxicity and X-irradiation-induced bone marrow injury were due to an induction of MT.^{34,35)} In the present study, Zn pretreatment suppressed the mutagenicity of B[a]P in wild-type mice but not in MT-I/II null mice. These results clearly indicate, therefore, that the protective effect of Zn pretreatment on the mutagenicity of B[a]P is also due to MT synthesis.

BPDE is well known to be the most effective carcinogen among the metabolites of B[a]P in animals and human.¹⁵⁾ Moreover, B[a]P produces various quinone derivatives and ROS, which are mutagenic and carcinogenic.³⁶⁾ Zn is known to induce the synthesis of MT through the activation of a metal-response elements that is located in the 5'-upstream region of the gene for MT.³⁷⁾ Some researchers have proposed that MT efficiently eliminates ROS, and that the rate of reaction of MT with hydroxyl radicals is several hundred-fold higher than that of GSH.³⁾ Many studies have demonstrated that Zn supplementation provides effective protection against various oxidative injuries.⁶⁾ Furthermore, the protective effect of MT against DNA damage caused by oxidative stress-inducing substances, such as hydrogen peroxide and cadmium, has been reported.³⁸⁾ In the present study, Zn significantly increased the MT levels in hepatic and bone marrow cells without any effects on GSH levels or the activities of other cellular antioxidative enzymes such as catalase and SOD. Miura *et al.* reported that GSH levels and catalase and SOD activities in bone marrow cells were not affected by treatment with bismuth compounds.³⁹⁾ Conrad *et al.* reported that neither the hepatic GSH levels nor catalase or SOD activities of Zn compound-treated MT-I/II null mice and wild-type mice were changed.⁴⁰⁾ Our results in the present study suggest that Zn-induced MT in bone marrow cells prevents B[a]P-induced oxidative DNA damage by scavenging free radicals.

Our previous report showed that MT-I/II null mice were more susceptible to B[a]P-induced DNA damage, without CYP1A activity being affected,

compared with wild-type mice.¹⁷⁾ In the present study, Zn pretreatment increased MT synthesis in the liver, the major metabolic organ of B[a]P, and also in bone marrow cells that are B[a]P-caused MN; however, other cellular antioxidants levels and metabolic activation of B[a]P were unchanged.

In conclusion, we found that the Zn pretreatment effectively prevented the mutagenicity of B[a]P, and its effect was due to MT induced by Zn. These results suggest that MT is a major preventive factor against the mutagenicity caused by B[a]P.

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