

Determination of Cyanide and Thiocyanate in Sugihiratake Mushroom Using HPLC Method with Fluorometric Detection

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A novel type of encephalopathy occurred in patients with chronic kidney diseases, which was associated with the ingestion of the Sugihiratake mushroom during the fall of 2004 in Japan. We attempted to investigate whether cyanide and thiocyanate are present in the Sugihiratake samples and determined the cyanide and thiocyanate levels in fifteen samples collected from different Japanese districts using HPLC with fluorometric detection. The cyanide ions and thiocyanate ions were detected in the ranges of N.D.–114.0 and N.D.–17.0 $\mu\text{g/g}$ in the samples, respectively. This is the first study to quantitatively detect cyanide and thiocyanate in the Sugihiratake mushrooms. This result demonstrated that cyanide exposure could occur from the intake of Sugihiratake mushrooms in one's diet. Furthermore, we discussed the possible association between cyanide and the onset of encephalopathy.

Key words — Sugihiratake, cyanide, thiocyanate, HPLC, encephalopathy

INTRODUCTION

Sugihiratake is the fungus *Pleurocybella porrigens*, which is a flat mushroom that grows on cedar and pine trees during the fall season, not only in the districts of northern Japan, but is also widely distributed in Japan.¹⁾ It has a specific flavor, and many Japanese have been favorably consuming it in the processed foods of the highly popular miso (fermented bean paste soup) and the deep-fried food tempura. However, during the fall of 2004 in Japan,

an outbreak of serious encephalopathy exclusively occurred in patients with chronic kidney diseases after the intake of this mushroom in many areas of Japan including the Akita, Yamagata, and Niigata Prefectures. Therefore, there have been some reports based on the clinical findings that encephalopathy was induced after the ingestion of this mushroom. The exact factors that induced the encephalopathy remain unclear and the association between the Sugihiratake mushroom intake and the onset of this novel type of encephalopathy is still currently controversial.

In the present study, we attempted to investigate the cyanide contents in wild Sugihiratake collected from several districts in Japan using a specific HPLC method with fluorometric detection, and were the first to detect cyanide in some of these samples. In addition, we discussed the possible association between cyanide intake and the onset of encephalopathy.

MATERIALS AND METHODS

Materials — The Sugihiratake mushroom samples were collected from the local health environment centers and the prefectural institutes of the public health and environmental science in Japan through the Ministry of Health, Labor and Welfare (MHLW) of Japan.

Reagents — A standard solution of potassium cyanide (0.1 M) was prepared by dissolving potassium cyanide (Wako Pure Chemicals, Osaka, Japan) in 0.1 M sodium hydroxide; the concentration of cyanide was calibrated by titration with silver nitrate using potassium iodide as the indicator according to the Liebig-Dènigès method.^{2,3)} A standard solution of potassium thiocyanate (Wako Pure Chemicals) was prepared using redistilled water. All other

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Table 1. Cyanide and Thiocyanate Contents of the Sugihiratake Mushroom Samples

Sample No.	Producing district	CN ⁻	SCN ⁻
		($\mu\text{g/g}$ dry weight)	($\mu\text{g/g}$ dry weight)
1	Akita 1	12.7	1.1
2	Akita 2	25.5	4.6
3	Yamagata	0.7	0.2
4	Niigata 1	1.8	0.5
5	Niigata 2	56.2	17.0
6	Mie	3.1	1.6
7	Gifu 1	22.1	10.3
8	Gifu 2	114.0	9.4
9	Fukui 1	3.0	1.1
10	Fukui 2	0.9	1.4
11	Ishikawa	N.D.	0.1
12	Kyoto	1.2	0.2
13	Ibaraki	96.6	8.4
14	Fukushima 1	0.6	0.1
15	Fukushima 2	0.3	N.D.

N.D.: not detected.

chemicals were of analytical reagent grade.

Preparation of Sample Solution — The freeze-dried Sugihiratake sample was ground to a fine powder using a grinder (Retsch GmbH, Haan, Germany), and a 500 mg test sample was extracted with 10.0 ml of 0.1 M sodium hydroxide by shaking overnight in a 50 ml centrifuge tube. A one ml portion was then placed in the outer well of the Conway cell and 1.0 ml of 0.1 M sodium hydroxide was placed in the center chamber. The Conway cell and ground-glass cover were coated with silicone grease, and a glass cover was placed on top of the microdiffusion cell, leaving a small space for the addition of the acidic solution. To the samples described above for the determination of cyanide, 1.0 ml of 1.2 M sulfuric acid was added to the outer chamber. Subsequently, the ground-glass cover was moved to seal the microdiffusion cell. These cells were carefully rotated in order to mix the solution in the outer chamber. The cells were then rotated every 30 min. Cyanide in the sample was allowed to diffuse for 4 hr at room temperature and the liberated hydrogen cyanide was absorbed into the sodium hydroxide solution in the center chamber. An aliquot from the center chamber solution was analyzed by HPLC.⁴⁾ For the recovery of thiocyanate from each sample during the pretreatment procedure, the Conway microdiffusion cell was kept for 24 hr and the collected thiocyanate in the center chamber was analyzed by HPLC. Using the pretreatment procedure described above, the spiked standard cyanide and

thiocyanate at the 1 μmol level were recovered at 100.2 ± 3.2 and $95.4 \pm 5.5\%$, respectively.

HPLC conditions⁴⁾ — The HPLC system consisted of a double-plunger pump (PU-1580, Jasco, Tokyo, Japan), an intelligent fluorescence detector (FP-920S, Jasco) with a xenon lamp and 12- μl flow cell, a chromato-integrator (D-2500, Hitachi, Tokyo, Japan) and a sample injector (7725i, Reodyne, CA, U.S.A.). The HPLC conditions were as follows: column, a strong base anion exchange resin, TSK-Gel SAX (150 \times 6 mm i.d., Tosoh Co., Tokyo, Japan); eluent, 0.1 M sodium acetate buffer (pH 5.0) containing 0.2 M sodium perchlorate (flow-rate, 1.0 ml/min); chlorination reagent, 0.1% chloramines T aqueous solution (flow-rate, 0.5 ml/min); pyridine-barbituric acid reagent, a mixture of barbituric acid (1.5 g), pyridine (15 ml), concentrated hydrochloric acid (3 ml) and redistilled water (82 ml) (flow-rate, 0.5 ml/min); the excitation and emission wavelengths of the detector were 583 and 607 nm, respectively.

RESULTS AND DISCUSSION

We attempted to investigate the cyanide content in the Sugihiratake samples and determined the cyanide in fifteen samples collected from different Japanese districts using HPLC with fluorometric detection. As shown in Table 1, we detected the cyanide ions and thiocyanate ions in the ranges of N.D.–

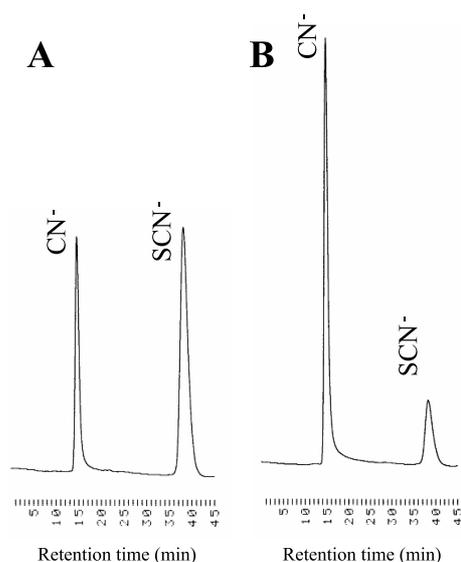


Fig. 1. Typical HPLC Chromatograms of Cyanide Ion and Thiocyanate Ion in a Sugihiratake Mushroom Sample A; CN⁻ and SCN⁻ Standard (10 mM, respectively), B; Sample 5 (Niigata).

114.0 $\mu\text{g/g}$ dry weight and N.D.–17.0 $\mu\text{g/g}$ dry weight, respectively. These levels would not be lethal doses for acute toxicity even if 1 kg of the maximum level sample was consumed, because the lethal dose of cyanide is estimated to be 200–300 mg for an adult human. This result demonstrates that cyanide and thiocyanate exposures would occur from the intake of Sugihiratake mushrooms.

As for the determination of cyanide, the conventional spectrophotometric method has been previously used in forensic toxicology and waste water regulation.⁵⁾ However, it is known that thiocyanate could cause a serious false positive using the conventional method. Therefore, even forensic scientists have often mistakenly estimated cyanide because of the false positive caused by thiocyanate.⁶⁾ In the present study, we used the specific and sensitive HPLC method of Toida *et al.*⁴⁾ after the pretreatment using the Conway cell. As shown in Fig. 1, we could simultaneously detect cyanide and thiocyanate using the HPLC system with an ion chromatographic column with non-interference determination. To our knowledge, this is the first report to accurately determine the content of cyanide and thiocyanate in Sugihiratake mushrooms. Since we confirmed the non-production of cyanide from linamarin, glucosidic cyanogens, using alkali solution (data not shown), we consider that the cyanide detected in the Sugihiratake samples would be present in the sodium form or potassium form.

Many food plants including agriculturally important species, such as cassava, flax, sorghum, alfalfa, peaches, almonds, and beans, are known to be cyanogenic.⁷⁾ Center African cassava flour contains sufficient quantities of cyanogens. When cassava is the staple part of the diet, the human daily consumption is equivalent to about one-half the lethal dose, which probably is thought to be the reason for the widespread and chronic neurological disorders called “konzo” found in this area.⁷⁾ In addition, the cyanide production has been observed in a wide range of fungi, such as *Phaeolepiota aurea*, *Rozites caperatus*, *Leucopaxillus giganteus*, and *Pleurocybella porringens* (Sugihiratake),⁸⁾ although there are no reports to describe the cyanide content in these fungi. Some reports suggested that the cyanide production in fungi could be associated with snow mold disease and fairy ring disease in some plants.⁹⁾

To date, some clinical case studies involving the outbreak of acute encephalopathy that occurred in Japan during the fall of 2004 have already been reported.^{10–13)} All the cases were involved with the intake of Sugihiratake and the patients had varying degrees of renal dysfunction. The common clinical syndrome was characterized by weakness and involuntary movements of the extremities or dysarthria at the onset of the disease and subsequent intractable focal motor seizures, resulting in the generalized status of epilepticus or a comatose state. Some brain MRI examinations revealed that diffuse lesions in the basal ganglia and multiple ringed lesions in the cerebral cortex.

While there are some studies that cyanide could induce encephalopathy.^{14–19)} Smith *et al.* showed that comparatively small doses of cyanide given over long intervals can produce histological changes in the central nervous system of the rat.¹⁴⁾ As for the clinical study, Rachinger *et al.* showed that the toxicity of cyanide caused cerebral damage, primarily to the basal ganglia in the case report of patients that attempted suicide with cyanide.¹⁷⁾ This symptom appears to be consistent with those cases that occurred in the Akita Prefecture of Japan.

Furthermore, a recent study showed that cyanide and thiocyanate do accumulate in haemodialysis patients due to tobacco smoking.²⁰⁾ Cyanide is known to be metabolized to thiocyanate by the enzyme rhodanese. This reaction is essential to life through its detoxification of cyanide, and thiocyanate synthesis can be accelerated under cyanide-loaded conditions such as tobacco smoking.²¹⁾ In addition, a

recent study showed that the risk of cerebral infarction stroke was significantly increased in individuals having high serum thiocyanate concentrations.²²⁾ There also is some evidence that thiocyanate enhances the action of glutamate in a subclass of neuronal glutamate receptors which are involved in the neurodegenerative disorders.²³⁾

These results suggested that the ingestion of cyanide from foods could also induce the accumulation of cyanide and thiocyanate in the blood of patients with chronic kidney diseases and might be associated with the onset of encephalopathy. However, since the human capacity for detoxification and the toxicity of cyanide and thiocyanate from food exposure in haemodialysis patients have not been fully investigated, a further investigation is necessary in order to elucidate the relationship between the cyanide in Sugihiratake mushrooms and the onset of a new type of encephalopathy that occurred in Japan during the fall of 2004. Nevertheless, it should be noted that there could still be other factors and other substances that caused this novel type of encephalopathy in addition to the ingestion of cyanide from Sugihiratake mushrooms.

In conclusion, we were the first to determine the cyanide content in the Sugihiratake samples collected from certain areas of Japan during the fall of 2004. In addition, we showed that some samples could contain cyanide in the range of N.D.–114.0 $\mu\text{g/g}$, and suggested that its form in the Sugihiratake samples could be the sodium or potassium salt. This finding suggested that cyanide in Sugihiratake might be associated with the onset of a novel type of encephalopathy in patients with chronic kidney diseases, which occurred in Japan during the fall of 2004.

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