

Contribution of Glutathione Peroxidase and Nitric Oxide to Potassium Bromate-Induced Oxidative Stress and Kidney Damage in Mice

Satoshi Watanabe,*^a Yoshihiro Yoshimura,^b and Tetsuya Fukui^a

^aDepartment of Health Chemistry, ^bDepartment of Analytical Chemistry, Faculty of Pharmaceutical Sciences, Hoshi University 4–41, Ebara 2-chome, Shinagawa-ku, Tokyo 142–8501, Japan

(Received September 6, 2001; Accepted September 21, 2001)

In order to confirm the participation of peroxynitrite in potassium bromate (KBrO₃)-induced oxidative stress and kidney damage in mice, we investigated effects of administration of nitric oxide (NO) synthase inhibitor on them. Cytoplasmic glutathione peroxidase (GPx) activity remarkably decreased within 3 hr after KBrO₃ administration, and then oxidative stress started to occur. However, kidney damage occurred 24 hr after KBrO₃ administration. Pre-administered *N*^G-monomethyl-L-arginine (L-NMMA), a NO synthase inhibitor, suppressed KBrO₃-induced oxidative stress and kidney damage. However, no effect of L-NMMA was observed on the KBrO₃-induced reduction of cytoplasmic GPx activity. These results suggest that reduction of cytoplasmic GPx activity resulted from the KBrO₃ administration initiates oxidative stress and that NO also participates in the promotion of KBrO₃-induced oxidative stress and kidney damage.

Key words — potassium bromate, oxidative stress, glutathione peroxidase, superoxide, kidney damage

INTRODUCTION

Potassium bromate (KBrO₃) is added to flour as a maturing agent, to dough and fish paste as a conditioner, and also added to beer or cheese.¹⁾ KBrO₃ has also been used as a constituent in cold-wave hair solutions.^{1,2)} In addition, bromate is generated as a disinfection by-product from bromide in the process of ozonation and chlorination of raw water.^{2,3)} Ingestion of KBrO₃ has been reported to induce renal oxidative stress in human and other mammals,^{1–4)} which is known to cause renal failure, methemoglobinemia and kidney cancer. This carcinogenesis has been suggested to be associated with oxidative DNA damage, which was induced by BrO₃[–] in the presence of intracellular sulfhydryl compound such as reduced glutathione (GSH) or cysteine.^{1,3,4)} We previously reported that *in vitro* lipid peroxidation by KBrO₃ progressed in the presence of GSH but not the oxidized form.⁵⁾ These results suggested that reactive species responsible for the induction of renal oxidative stress were generated from KBrO₃. How-

ever, the reactive species in the KBrO₃-induced oxidative stress are not clear yet. In the present study, we have demonstrated the *in vitro* generation of superoxide anion radical (O₂[–]) from KBrO₃ using electron spin resonance (ESR) spectrometry, and have shown the participation of peroxynitrite (ONOO[–]) possibly produced from O₂[–] and physiological nitric oxide (NO) in the KBrO₃-induced renal oxidative stress and kidney damage by administering NO synthase inhibitor to mice.

MATERIALS AND METHODS

Chemicals — Glutathione reductase from bakers yeast (EC 1.6.4.2), xanthine oxidase from milk (EC 1.1.3.22), GSH and oxidized glutathione, xanthine, and bovine serum albumin were obtained from Sigma Chemical Co. (St. Louis, MO, U.S.A.). 2,4-Dinitrophenylhydrazine was purchased from Tokyo Chemical Industry Co. (Tokyo, Japan). 5,5-Dimethyl-1-pyrroline *N*-oxide (DMPO) was obtained from Dojindo Laboratories (Kumamoto, Japan). KBrO₃ and *N*^G-monomethyl-L-arginine acetate (L-NMMA) were purchased from Wako Pure Chemical Industries (Osaka, Japan). Other chemicals of the highest grade were obtained commercially.

*To whom correspondence should be addressed: Department of Health Chemistry, Faculty of Pharmaceutical Sciences, Hoshi University 4–41, Ebara 2-chome, Shinagawa-ku, Tokyo 142–8501, Japan. Tel.: +81-3-5498-5773; Fax: +81-3-5498-5771; E-mail: satoshi@hoshi.ac.jp

ESR Spectrometry — ESR spectra were recorded on a JEOL JES-RE1X spectrometer using a flat quartz cell designed for aqueous solution. Conditions of ESR spectrometry were as follows : magnetic field, 336.3 ± 5 mT; power, 8.0 mT; modulation frequency, 100 kHz; frequency, 9.425 GHz; modulation amplitude, 0.063 mT; gain, 500; time scan, 1 min; time constant, 0.03 s.

Animals — Five-week-old male ddY mice purchased from Tokyo Experimental Animal Supply Co. (Tokyo, Japan) were given MF pellet basal diet (Oriental Yeast Co., Tokyo, Japan) and tap water freely, and were used after 1 week of acclimation. Mice were housed in an air-conditioned room with temperature of $23 \pm 1^\circ\text{C}$, a humidity of $50 \pm 3\%$, and a 12 hr light and dark cycle.

Treatment of Animals — In the time-course experiments on the oxidative stress status caused by KBrO_3 , mice received a single i.p. administration of KBrO_3 dissolved in 0.9% NaCl at a dose of 200 mg/kg and were sacrificed at 0, 3, 6, 9 and 24 hr after treatment. The experiments for effects of L-NMMA on the KBrO_3 -induced oxidative stress and kidney damage were carried out by the injection of 20 mg/kg or 40 mg/kg of L-NMMA dissolved in DMSO or DMSO alone to the abdominal cavity of mice at 30 min before KBrO_3 administration. The mice were sacrificed 6 hr after the KBrO_3 administration. In both experiments, kidneys were immediately excised, weighed and used for the measurement of protein carbonyl, thiobarbituric acid-reactive substances (TBARS), GSH and antioxidative enzyme activities. Whole blood was removed and used for the measurement of serum creatinine level. This experimental design was approved by the Animal Experiment Committee of Hoshi University and the mice were cared for in accordance with the Guidelines Concerning the Care and Use of Laboratory Animals.

Measurement of Serum Creatinine Level — Serum creatinine level was measured with Creatinine Test-Wako (Wako Pure Chemical Industries, Japan).

Measurement of Protein Carbonyl Group — Protein carbonyl group in kidney was assayed by the method of Reznick *et al.*⁶ Absorbance at 370 nm of guanidine hydrochloride solution was measured using Hitachi U-2000 Spectrophotometer (Hitachi Co., Tokyo, Japan). Carbonyl concentration was calculated as follows, C (μM) = $\text{OD}_{370} \times 45.45$.⁶

Measurement of TBARS Content — TBARS content in kidney was measured by the method of Ohkawa *et al.*⁷ Absorbance at 535 nm of dehydrated

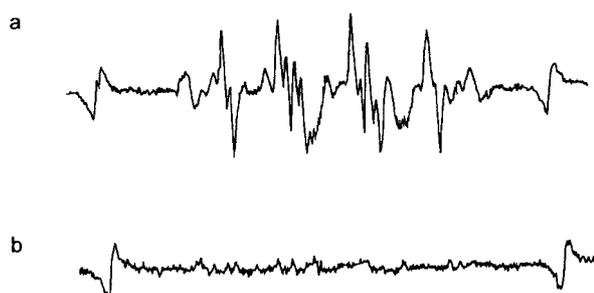


Fig. 1. ESR Signal of DMPO-OOH Adduct Produced from KBrO_3 in the Presence of GSH and Transition Metal

The reaction mixture contained 1 mM KBrO_3 , 1 mM GSH, 0.1 mM FeSO_4 and 100 mM DMPO (a), 1 mM oxidized glutathione instead of 1 mM GSH was added to the reaction mixture (b). ESR conditions were described in "MATERIALS AND METHODS."

organic layer was measured and the molar extinction coefficient of malondialdehyde ($\epsilon = 1.56 \times 10^2 \text{ mM}^{-1} \text{ cm}^{-1}$)⁷ was used to calculate TBARS concentration.

Measurement of GSH Content — GSH content in kidney was determined as described in the previous report⁸ using HPLC-UV detector system.

Measurement of Antioxidative Enzyme Activities — Copper/zinc-containing superoxide dismutase (Cu/Zn-SOD), catalase and glutathione peroxidase (GPx) activities were measured spectroscopically as described previously.⁸

Measurement of Protein Concentration — Protein concentration was determined by the method of Lowry *et al.*,⁹ using bovine serum albumin as the standard protein.

Statistics — Data were expressed as the mean \pm S.D. One-way ANOVA was used to determine any significant differences ($p < 0.05$) between means. When significant differences were found, Duncan's multiple range test was used to determine the exact nature of the difference.

RESULTS

In Vitro O_2^- Generation from KBrO_3

Reactive species generated from KBrO_3 in the presence of sulfhydryl group were identified by ESR spectrometry. ESR signal of O_2^- adduct to DMPO (DMPO-OOH) was detected in the presence of KBrO_3 , GSH and transition metal (Fig. 1a). While, no DMPO-OOH signal was observed when oxidized glutathione instead of GSH was added to the reaction mixture (Fig. 1b).

Table 1. Changes of Parameters for Oxidative Stress and Kidney Damage after KBrO₃ Administration to Mice

Time (h)	BW (g)	KW (g)	RKW	TBARS	Carbonyl	GSH	GPx	Catalase	Cu/Zn-SOD
				($\mu\text{mol/g tissue}$)	(nmol/g tissue)	(nmol/g tissue)			
0	31.6 \pm 1.4 ^{a)}	0.51 \pm 0.02 ^{a)}	1.63 \pm 0.09 ^{a)}	2.56 \pm 0.68 ^{a)}	99.9 \pm 27.3 ^{a)}	85.0 \pm 7.2 ^{a)}	0.39 \pm 0.07 ^{a)}	153 \pm 23 ^{a)}	15.0 \pm 1.9
3	32.3 \pm 0.5 ^{a)}	0.56 \pm 0.04 ^{a, b)}	1.73 \pm 0.15 ^{a)}	4.09 \pm 0.92 ^{b)}	163.8 \pm 39.3 ^{a, b)}	74.6 \pm 6.3 ^{b)}	0.14 \pm 0.04 ^{b)}	116 \pm 11 ^{b)}	12.3 \pm 2.4
6	32.3 \pm 2.0 ^{a)}	0.56 \pm 0.04 ^{a, b)}	1.74 \pm 0.23 ^{a)}	7.26 \pm 0.40 ^{c)}	245.4 \pm 38.7 ^{b)}	60.4 \pm 8.2 ^{c)}	0.12 \pm 0.02 ^{b)}	92 \pm 15 ^{c)}	16.3 \pm 1.8
9	29.0 \pm 1.0 ^{b)}	0.55 \pm 0.08 ^{a)}	1.90 \pm 0.21 ^{a)}	3.29 \pm 1.27 ^{a, b)}	362.4 \pm 83.4 ^{c)}	54.3 \pm 4.6 ^{c)}	0.13 \pm 0.02 ^{b)}	84 \pm 9 ^{c)}	15.1 \pm 3.0
24	28.2 \pm 0.6 ^{b)}	0.63 \pm 0.04 ^{b)}	2.23 \pm 0.19 ^{b)}	2.99 \pm 0.41 ^{a, b)}	420.0 \pm 74.7 ^{c)}	52.3 \pm 4.3 ^{c)}	0.11 \pm 0.03 ^{b)}	70 \pm 10 ^{c)}	14.4 \pm 2.9

KBrO₃ (200 mg/kg) dissolved in 0.9% NaCl was administered intraperitoneally to male ddY mice. Each parameter was measured as described in "MATERIALS AND METHODS." BW, KW and RKW indicate body weight, kidney weight and relative kidney weight, respectively. RKW was calculated as follows, KW \times 100/BW. Values are mean \pm S.D. ($n = 4$). a, b, c) Values not sharing a common letter are significantly different at $p < 0.05$ (ANOVA with Duncan's multiple-range test).

Table 2. Effect of L-NMMA Administration on the KBrO₃-Induced Oxidative Stress and Kidney Damage in Mice

Group	BW (g)	KW (g)	RKW	Creatinine	TBARS	Carbonyl	GSH	GPx	Catalase	Cu/Zn-SOD
				(mg/l serum)	($\mu\text{mol/g tissue}$)	(nmol/g tissue)	(nmol/g tissue)			
Vehicle	30.8 \pm 1.5	0.45 \pm 0.02	1.46 \pm 0.12 ^{a)}	3.06 \pm 0.69 ^{a)}	1.78 \pm 0.18 ^{a)}	98 \pm 15 ^{a)}	83.7 \pm 6.3 ^{a)}	0.48 \pm 0.04 ^{a)}	154 \pm 10 ^{a)}	17.8 \pm 2.4
KBrO ₃ 200 mg/kg	29.6 \pm 0.6	0.50 \pm 0.04	1.87 \pm 0.15 ^{b)}	10.22 \pm 2.36 ^{b)}	2.38 \pm 0.08 ^{b)}	228 \pm 21 ^{b)}	57.7 \pm 1.6 ^{b)}	0.07 \pm 0.01 ^{b, c)}	70 \pm 17 ^{b, d)}	18.9 \pm 2.7
KBrO ₃ 200 mg/kg + L-NMMA 20 mg/kg	29.2 \pm 2.1	0.46 \pm 0.02	1.58 \pm 0.08 ^{a, c)}	6.47 \pm 1.71 ^{c)}	2.36 \pm 0.06 ^{b)}	242 \pm 31 ^{b)}	68.0 \pm 2.6 ^{c)}	0.12 \pm 0.02 ^{b)}	52 \pm 15 ^{d)}	14.5 \pm 4.4
KBrO ₃ 200 mg/kg + L-NMMA 40 mg/kg	29.2 \pm 0.8	0.50 \pm 0.03	1.69 \pm 0.06 ^{c)}	4.43 \pm 2.58 ^{a, c)}	2.06 \pm 0.06 ^{c)}	125 \pm 26 ^{a, c)}	80.5 \pm 8.8 ^{a)}	0.05 \pm 0.02 ^{c)}	88 \pm 15 ^{b, c)}	16.2 \pm 3.4
L-NMMA 40 mg/kg	30.3 \pm 1.8	0.47 \pm 0.04	1.56 \pm 0.06 ^{a, c)}	2.72 \pm 1.11 ^{a)}	2.12 \pm 0.12 ^{c)}	141 \pm 10 ^{c)}	82.3 \pm 6.6 ^{a)}	0.46 \pm 0.07 ^{a)}	105 \pm 19 ^{c)}	17.3 \pm 3.6

KBrO₃ dissolved in 0.9% NaCl was administered intraperitoneally to ddY male mice at 30 min after L-NMMA administration. L-NMMA dissolved in DMSO was administered intraperitoneally to mice as low dose (20 mg/kg) and high dose (40 mg/kg) at 30 min before KBrO₃ administration. All mice were sacrificed 6 h after KBrO₃ administration. Each parameter was measured as described in "MATERIALS AND METHODS." BW, KW and RKW are the same in Table 1. Values are mean \pm S.D. ($n = 4$). a, b, c, d) Values not sharing a common letter are significantly different at $p < 0.05$ (ANOVA with Duncan's multiple-range test).

Time-Course of Oxidative Stress Status after KBrO₃ Administration

Effect of administration of KBrO₃ to mice on antioxidative enzyme activities which are known to be influenced by O₂⁻ and/or other reactive species and on kidney damage was investigated. Cytoplasmic catalase and GPx activities were remarkably decreased by the KBrO₃ administration. In particular, GPx activity reached near the lowest level within 3 hr after the treatment (Table 1). Protein carbonyl group, which is known to be an important parameter to discriminate the oxidative stress status, gradually elevated, and the highest level was observed at 24 hr after the treatment under the experimental condition (Table 1). However, renal TBARS started to increase within 3 hr and reached the maximum level at 6 hr after the treatment, and then gradually decreased (Table 1). Renal GSH slowly decreased, and the lowest level was observed 24 hr after the treatment (Table 1). Kidney weight and relative kidney weight, which are parameters of kidney damage, significantly increased at 24 hr after the treatment (Table 1).

Effect of L-NMMA on the KBrO₃-Induced Oxidative Stress and Kidney Damage

Next, in order to confirm the involvement of NO and ONOO⁻ in the KBrO₃-induced oxidative stress, L-NMMA, a NOS inhibitor, was pre-administered to mice. As shown in Table 2, L-NMMA treatment alone caused the increases in TBARS and protein carbonyl group, and the decrease in catalase activity. In other parameters measured in the experiment, no significant effect of L-NMMA alone was observed (Table 2). In KBrO₃-treated mice, the decreased cytoplasmic catalase activity was recovered by 40 mg/kg L-NMMA to nearly the same level as that obtained by L-NMMA alone. However, no effect of L-NMMA on the KBrO₃-induced decrease of cytoplasmic GPx activity was observed under the experimental condition (Table 2). Increases in protein carbonyl group and TBARS, parameters of oxidative stress, were prevented in L-NMMA-treated mice (Table 2). L-NMMA also prevented the KBrO₃-induced decrease in GSH and increases in relative kidney weight and serum creatinine level (Table 2).

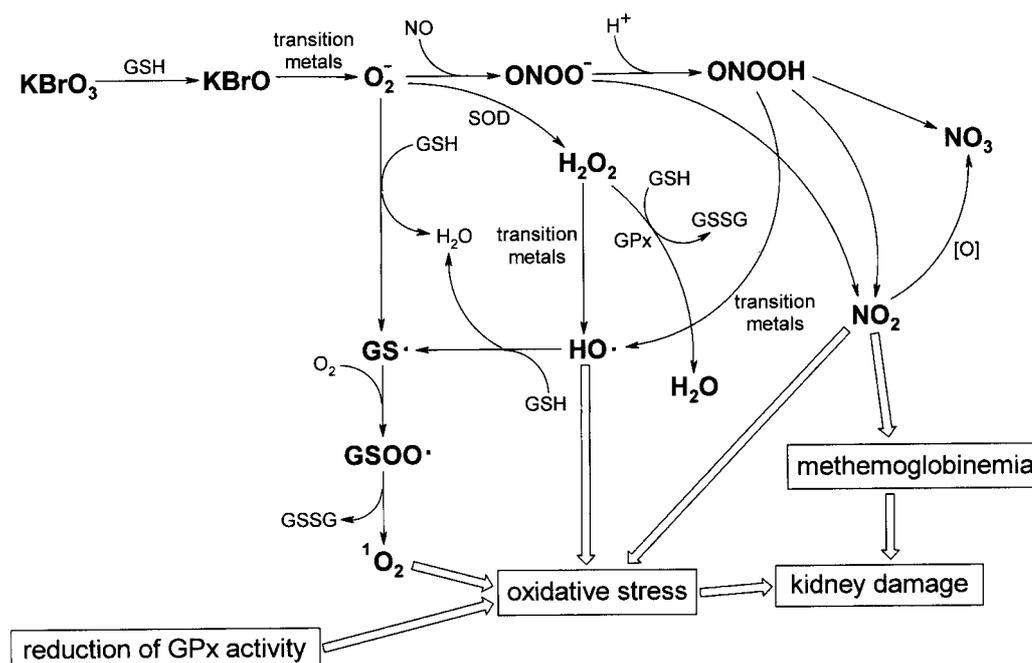


Fig. 2. Possible Pathways for the KBrO_3 -Induced Generation of Reactive Species which Cause Oxidative Stress, Methemoglobinemia and Kidney Damage in Mouse

DISCUSSION

In the oxidative stress status, increase in TBARS and decrease in GSH in various tissues have been generally observed.¹⁰ TBARS and protein carbonyl group are useful markers for the detection of cellular damage caused by reactive oxygen species (ROS) if they are taken together with other markers such as cellular concentrations of various antioxidants.^{11,12} Our results of the time-course of TBARS suggest that renal oxidative stress is induced within 6 hr after KBrO_3 administration. In addition, the level of oxidatively-modified protein in kidney also increased remarkably. Elevation of the level of oxidatively-modified protein has been observed in several oxidative stress-related diseases, such as Alzheimer's disease¹³ and Parkinson's disease.¹⁴

Cu/Zn-SOD, catalase and GPx are known as antioxidative enzymes in various tissues which physiologically suppress the oxidative stress by catalyzing the removal of ROS. In our experiment, cytoplasmic catalase and GPx activities were distinctly reduced within 3 hr after KBrO_3 administration. Therefore, the KBrO_3 -induced oxidative stress status was likely to have resulted from the reductions of cytoplasmic catalase and GPx activities.¹⁵ That is, the reductions of cytoplasmic catalase and GPx activities caused the elevation of cellular hydrogen peroxide, which is mainly removed by these en-

zymes.

Relative tissue weight is a marker which generally elevates when tissue is damaged.¹⁶ Since relative kidney weight elevated after the increases in renal TBARS and protein carbonyl group, KBrO_3 -induced kidney damage appears to occur following oxidative stress. Intraperitoneally-injected L-NMMA, an inhibitor of NOS, is known to cause decrease in NO generation in liver and kidney by nonselectively inhibiting the constitutive and inducible NOS activities.¹⁷ Since KBrO_3 -induced increases in relative kidney weight and serum creatinine levels were significantly suppressed by L-NMMA, NO seemed to be involved in the KBrO_3 -induced kidney damage.

Pre-administration of L-NMMA significantly prevented the KBrO_3 -induced elevations of renal TBARS and protein carbonyl group. Furthermore, KBrO_3 -induced reduction of renal GSH content was also significantly prevented by L-NMMA. These results suggest that L-NMMA exhibits the suppressive effect on the oxidative stress status caused by KBrO_3 . Since L-NMMA is not a free radical scavenger,¹⁸ suppressive effect of L-NMMA on the oxidative stress is likely to have resulted from the reduction of NO generation. NO is known to rapidly react with O_2^- to form ONOO^- , which is a powerful oxidant and has cytotoxicity.¹⁹ Although being a result of *in vitro* experiment, O_2^- generation from KBrO_3 in the

presence of GSH but not oxidized glutathione indicates that one of the causes of KBrO_3 -induced oxidative stress is the elevation of the level of cellular ONOO^- produced from O_2^- and NO.

Reduction of not cytoplasmic GPx activity but cytoplasmic catalase activity in KBrO_3 -treated mice was significantly prevented by L-NMMA administration. However, catalase is known to possess much lower hydrogen peroxide-removing ability than GPx in the cytoplasmic fraction of the cell.²⁰ Therefore, it is possible that incomplete suppression by L-NMMA of KBrO_3 -induced oxidative stress was mainly resulted from its inability to recover the cytoplasmic GPx activity. It has been reported that element selenium in the active site of GPx nonenzymatically reacts with ONOO^- to bring about the inactivation of the enzyme^{15,20} However, since L-NMMA had no influence on the reduction of cytoplasmic GPx activity caused by KBrO_3 administration, it is likely that O_2^- generated from KBrO_3 rather than NO is directly involved in the KBrO_3 -induced reduction of GPx activity.

KBrO_3 is also known to cause methemoglobinemia,^{1,3} and mechanism for its induction by KBrO_3 is not clear yet. According to our results, one of the mechanisms could be explained on the basis of the elevation of ONOO^- level in blood.²¹ Thus, tentative mechanisms of KBrO_3 -induced oxidative stress, kidney damage and methemoglobinemia are shown in Fig. 2. The pathway of the generation of singlet oxygen from KBrO_3 , previously proposed by us,⁵ is also shown in Fig. 2. In the recent paper, Rahman *et al.* proposed that NO acts as an antioxidant in KBrO_3 -induced renal oxidative stress.¹⁸ However, present results surely suggest that NO and ONOO^- act as accelerator on KBrO_3 -induced oxidative stress and kidney damage.

Acknowledgements The research was supported in part by the Ministry of Education, Science, Sports, and Culture of Japan.

REFERENCES

- 1) Chipman, J. K., Davies, J. E., Parsons, J. L., Nair, J., O'Neill, G. and Fawell, J. K. (1998) DNA oxidation by potassium bromate; a direct mechanism or linked to lipid peroxidation? *Toxicology*, **126**, 93–102.
- 2) Ueno, H., Oishi, K., Sayato, Y. and Nakamuro, K. (2000) Oxidative cell damage in Kat-Sod assay of oxyhalides as inorganic disinfection by-products and their occurrence by ozonation. *Arch. Environ. Contam. Toxicol.*, **38**, 1–6.
- 3) Parsons, J. L. and Chipman, J. K. (2000) The role of glutathione in DNA damage by potassium bromate. *Mutagenesis*, **15**, 311–316.
- 4) Sai, K., Uchiyama, S., Ohno, Y., Hasegawa, R. and Kurokawa, Y. (1992) Generation of active oxygen species in vitro by the interaction of potassium bromate with rat kidney cell. *Carcinogenesis*, **13**, 333–339.
- 5) Watanabe, S., Miyasaka, K., Kawana, K. and Kawauchi, S. (1993) Production of active oxygen species and lipid peroxidation by the reaction of potassium bromate and reduced glutathione. *Jpn. J. Toxicol. Environ. Health*, **39**, 445–452.
- 6) Reznick, A. Z., Kagan, V. E., Ramsey, R., Tsuchiya, M., Khwaja, S., Serbinova, E. A. and Packer, L. (1992) Antiradical effects in L-propionyl carnitine protection of the heart against ischemia-reperfusion injury: the possible role of iron chelation. *Arch. Biochem. Biophys.*, **296**, 394–401.
- 7) Ohkawa, H., Ohishi, N. and Yagi, K. (1979) Assay for lipid peroxides in animal tissues by thiobarbituric acid reaction. *Anal. Biochem.*, **95**, 351–358.
- 8) Togashi, S., Takahashi, N., Watanabe, S., Ishiguro, A. and Fukui, T. (2000) Suppressive effects of uracil, tyrosine, and phenylalanine contained in human-placenta extract on acute ethanol-induced liver injury in mice. *J. Health Sci.*, **46**, 126–131.
- 9) Lowry, O. H., Rosebrough, N. J., Farr, A. L. and Randall, R. J. (1951) Protein measurement with the Folin phenol reagent. *J. Biol. Chem.*, **193**, 265–275.
- 10) Imai, K., Aimoto, T., Sato, M. and Kimura, R. (1991) Antioxidative effect of protoporphyrin on lipid peroxidation in tissue homogenates of intravenously administrated rats. *J. Pharmacobio-Dyn.*, **14**, 20–24.
- 11) Davies, M. J., Fu, S., Wang, H. and Dean, R. T. (1999) Stable markers of oxidant damage to proteins and their application in the study of human disease. *Free Radic. Biol. Med.*, **27**, 1151–1163.
- 12) Pantke, U., Volk, T., Schmutzler, M., Kox, W. J., Sitte, N. and Grune, T. (1999) Oxidized proteins as a marker of oxidative stress during coronary heart surgery. *Free Radic. Biol. Med.*, **27**, 1080–1086.
- 13) Pocerlich, C. B., La Fontaine, M. and Butterfield, D. A. (2000) In-vivo glutathione elevation protects against hydroxyl free radical-induced protein oxidation in rat brain. *Neurochem. Int.*, **36**, 185–191.
- 14) Mizuno, Y., Ikebe, S., Hattori, N., Nakagawa-Hattori, Y., Mochizuki, H., Tanaka, M. and Ozawa, T. (1995) Role of mitochondria in the etiology and pathogenesis of Parkinson's disease. *Biochem. Biophys. Acta*, **1271**, 265–274.

- 15) Asahi, M., Fujii, J., Suzuki, K., Seo, H. G., Kuzuya, T., Hori, M., Tada, M., Fujii, S. and Taniguchi, N. (1995) Inactivation of glutathione peroxidase by nitric oxide. *J. Biol. Chem.*, **270**, 21035–21039.
- 16) Sai, K., Takagi, A., Umemura, T., Hasegawa, R. and Kurokawa, Y. (1991) Relation of 8-hydroxydeoxyguanosine formation in rat kidney to lipid peroxidation, glutathione level and relative organ weight after a single administration of potassium bromate. *Jpn. J. Cancer Res.*, **82**, 165–169.
- 17) Arteel, G. E., Briviba, K. and Sies, H. (1999) Protection against peroxynitrite. *FEBS Lett.*, **445**, 226–230.
- 18) Rahman, A., Ahmed, S., Khan, N., Sultana, S. and Athar, M. (1999) Glyceryl trinitrate, a nitric oxide donor, suppresses renal oxidant damage caused by potassium bromate. *Redox Rep.* **4**, 263–269.
- 19) Halliwell, B. and Gutteridge, J. M. C. (1999) *Free Radicals in Biology and Medicine 3rd edition*, Oxford University Press Inc., New York.
- 20) Sies, H., Sharov, V. S., Klotz, L.-O. and Briviba, K. (1997) Glutathione peroxidase protects against peroxynitrite-mediated oxidations. *J. Biol. Chem.*, **272**, 27812–27817.
- 21) Giri, U., Iqbal, M. and Athar, M. (1999) Potassium bromate (KBrO₃) induces renal proliferative response and damage by elaborating oxidative stress. *Cancer Lett.*, **135**, 181–188.